

# **Basic Concepts Of Chemistry 9th Edition**

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Concepts of Chemistry ... Class XI Chapter 1 - Some Basic Concepts of Chemistry  
Chemistry = 0.0767 g  
Since carbon and hydrogen are the only constituents of the compound, the total mass of the compound is: =  
 $0.9217 \text{ g} + 0.0767 \text{ g} = 0.9984 \text{ g}$   
Percent of C in the compound = 92.32%  
Percent of H in the compound = 7.68%  
Moles of carbon in the compound = 7.69  
Moles of hydrogen in the compound = = 7.68  
Ratio of carbon to ...  
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