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Chemistry Classifying Matter Answer Key

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Chemistry Classifying Matter Answer Key

Chemistry: Classifying Matter Name _____.

Classify each of the materials below. In the center column, state whether the material is a pure substance or a mixture. If the material is a pure substance, further classify it as either an element or compound in the right column. Similarly, if the material is a mixture, further classify it as homogeneous or heterogeneous in the right column.

Chemistry: Classifying Matter Name	Material	Pure Substance ...	Key Takeaways
			Key Points. Matter can exist in one of three main states: solid, liquid, or gas. Solid matter is composed of tightly packed particles. A solid will

Key

retain its shape; the particles are not free to move around. Liquid matter is made of more loosely packed particles. It will take the shape of its container. Classification of Matter | Boundless

Chemistry Chemistry: Classifying Matter 1. Define matter, pure, mixture, element, compound, solution and mechanical mixture.

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properties are characteristics that describe how matter changes its chemical structure or composition. 1.2: The Classification of Matter - Chemistry

LibreTexts Chemistry: Classifying Matter. Classify each of the materials below. In the center column, state whether the material is a . pure substance. or a . mixture. If the material is a pure substance, further classify it as either an . element. or . compound. in the right column. Similarly, if the material is a mixture, further classify it as . homogeneous . or . heterogeneous

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different type of matter Chemical Change A change in matter in which the substances that make up the matter change into other substances with new physical and chemical properties, they are often called chemical

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Classification Of Matter and

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Changes ... Matter can be divided into two main classifications—substances and mixtures. A substance is matter with a composition that is always the same. This means that a given substance is always made up of one or more atoms in the same combinations. Aluminum, oxygen, water, and sugar are examples of substances. Foundations of Chemistry The properties that chemists use to describe matter fall into two general categories. Physical properties are characteristics that describe matter. They include characteristics such as size, shape, color, and mass. These characteristics can be observed or measured without changing the identity of the matter in question. 1.2: The Classification

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