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central carbon in  $\text{CH}_3\text{C}\equiv\text{N}$  and the bond angle  $\text{CCN}$  are a.  $\text{sp}^2$ ,  $180^\circ$ . b.  $\text{sp}$ ,  $180^\circ$ . c.  $\text{sp}^2$ ,  $120^\circ$ . d.  $\text{sp}^3$ ,  $109^\circ$ .

2. Which of the following statements about an  $\text{sp}$  hybridized carbon is FALSE? a. It is divalent. b. It forms bonds that are linear. c. It has two p orbitals. d. General

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54.6cm<sup>3</sup> of benzene, vapour pressure of this solution, is found to be 98.88 mm Hg. calculate the freezing point of the solution..

Given: Enthalpy of vapourisation of benzene(l)=394.57 Jg<sup>-1</sup>. NEET

Chemistry Solutions Questions

Solved Organic Chemistry Question

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have one of two possible

compounds. Compound A, has a

melting point of 160°C, compound

B has a melting point of 161°C.

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