

Solutions Worksheet 1 Molarity Complete The Table Answers

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Complete The Table Answers

Solutions Worksheet 1 Molarity

Complete Solutions Worksheet #1:

Molarity 1) Complete the table:

Solute formula Molar Mass of solute (g/mole) Mass of Solute (g) Moles of solute (mole) Molarity of solution (M) Volume of solution (L)

a NaCl 1.0 1.0

b NaOH 117.0 4.0

c MgCl₂ 190.3 2.0

d NaCl 292.5 0.5

e KBr 238.0 2.0

f NaCl 1.5 1.0

g NaCl 1.0 0.25

Solutions Worksheet 1 Molarity
Complete The Table Solutions
Worksheet #1: Molarity 1)
Complete the table: Solute formula
Molar Mass of solute (g/mole) Mass
of Solute (g) Moles of solute (mole)

Molarity of solution (M) Volume of
solution (L)

a NaCl 1.0 1.0

b NaOH 117.0 4.0

c MgCl₂ 190.3 2.0

d NaCl 292.5 0.5

e KBr 238.0 2.0

f NaCl

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1.5 1.0 g NaCl 1.0 0.25 Solutions
Worksheet #1: Molarity Molar Mass
of Mass of ... Calculate molarity if
25.0 mL of 1.75 M HCl diluted to
65.0 mL. Calculate molarity by
dissolving 25.0g NaOH in 325 mL of
solution. Calculate grams of solute
needed to prepare 225 mL of 0.400
M KBr solution. Calculate mL of
0.650M KNO₃ needed to contain
25.0g KNO₃. Molarity 1

(Worksheet) Name: Date: Molarity
About Chemistry

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Complete the table for the following
aqueous solutions Name: Date:

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PDF What is the molarity of the following solutions given that: 1)

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1.0 moles of potassium fluoride is dissolved to make 0.10 L of solution. 2) 1.0 grams of potassium fluoride is dissolved to make 0.10 L of solution. 3) 1.0 grams of potassium fluoride is dissolved to make 0.10 mL of solution. 4) 952 grams of ammonium carbonate are dissolved to make 1750 mL of solution. Molarity Worksheet W

331 1. What mass of the following chemicals is needed to make the solutions indicated? a. 1.0 liter of a 1.0 M mercury (II) chloride (HgCl_2) solution. b. 2.0 liters of a 1.5 M sodium nitrate (NaNO_3) solution. c. 5.0 liters of a 0.1 M $\text{Ca}(\text{OH})_2$ solution. d. 100 mL of a 0.5 M $(\text{NH}_4)_3\text{PO}_4$ solution. 2. Calculate the molarity of the following solutions. a. Molarity of Solutions Acces PDF Solutions

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Molarity Name _____ Date _____

Period _____ Solutions Worksheet

#1: Molarity 1) Complete the table:

Solute formula Molar Mass of solute

(g/mole) Mass of Solute (g) Moles of

solute (mole) Molarity of solution

(M) Volume of solution (L) a NaCl

1.0 1.0 b NaOH 117.0 4.0 c MgCl 2

190.3 2.0 d NaCl 292.5 0 .5 e KBr

238.0 2.0 f NaCl 1.5 1.0 g NaCl 1.0

0.25 h ... molarconc.pdf - Name

Date Period Solutions Worksheet#1

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... Solutions Worksheet #1: Molarity

1) Complete the table: Solute

formula Molar Mass of solute

(g/mole) Mass of Solute (g) Moles of

solute (mole) Molarity of solution

(M) Volume of solution (L) a NaCl 10

10 b NaOH 1170 40 c MgCl₂ 1903

20 d NaCl 2925 0 5 [MOBI] Molarity

Molality Mass And Mole Fraction

Answers Molarity is the number of moles dissolved per liter of solution.

4:14 - How molecular mass relates to the mass of 1 mole of a

molecule: The mass of 1 mole of a substance in grams is equal to the molecular mass of the

substance. Molarity Problems

(Chemistry 1 Exam Solution

Breakdown) Calculate the molarity

if a flask contains 1.54 moles

potassium sulfate in 125 ml of

solution. $1.54 \text{ mol K}_2\text{SO}_4 = 12.3 \text{ M}$

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Complete The Table Answers

K_2SO_4 . 0.125 L soln. A chalice contains 36.45 grams ammonium chlorite in 2.36... Molarity

Worksheet 2 ANSWERS 1 L 1 mole.

2. Calculate the mass of $CoCl_3 \cdot 8H_2O$ required to prepare 300.0 mL of a 0.520 M solution. $0.300 \text{ L} \times 0.520 \text{ moles} \times 309.56 \text{ g} = 48.3 \text{ g}$. 1 L 1 mole. 3.

150.0 g of NaCl is dissolved in 250.0 mL of water, calculate the molarity. Molarity Worksheet # 1 6)

A solution is made by dissolving 1.0 g of potassium dihydrogen phosphate in 1.0 L of water.

Assume that the volume of the resulting solution is 1.0 L and that the density of water and the resulting solution is 1.0 g/mL.

Determine the molarity of the solution. 7) A student is instructed to prepare 3.00 L of a 0.100 M

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Table Question: Acids And Bases, Molarity, And Titration Part 1 Online Worksheet Solution 1: Preparation Of Potassium Hydrogen Phthalate - Acid Solution (include Units, Use Periodic Table On Blackboard To Calculate Molar Mass) # Of Sig. Figs. Given Given Mass Of KHP (Top Loading Balance) Mass Of KHP (Analytical Balance) 5.1 Fla Molar Mass, KHP 20 5 5 Moles, KHP LA ...

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