

# **Types Of Solutions Powerpoint**

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SOLUTE AND SOLVENT • THE  
SUBSTANCE WHICH IS DISSOLVED  
IN A LIQUID IS CALLED “SOLUTE”. •  
THE LIQUID IN WHICH SOLUTE IS...

3. SOLUTION □ A solution is a  
homogeneous mixture of two or  
more substances. □ Examples: salt  
solution, sugar solution,... 4.

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HOMOGENEOUS MIXTURE OF TWO  
OR MORE SUBSTANCES. A  
HOMOGENEOUS MIXTURE MEANS  
THAT THE MIXTURE IS JUST THE  
SAME THROUGHOUT SOME  
COMMON EXAMPLE OF SOLUTIONS  
ARE SALT SOLUTION, SUGAR  
ALLOYS AND AIR. Types Of Solution  
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of Solutions Depending upon the  
dissolution of the solute in the  
solvent, solutions can be  
categorized into supersaturated  
solution, unsaturated and saturated  
solutions. A supersaturated solution  
comprises of a large amount of  
solute at a temperature wherein it

will be reduced as a result the extra solute will crystallize quickly. Types of Solutions - Different Types, Homogeneous ... SOLUTIONS Chapter 13 Soap forms micelle which is the bubble. A thin film of water is coated on both surfaces by soap molecules. Hard water destroys soap s surfactant ... - A free PowerPoint PPT presentation (displayed as a Flash slide show) on PowerShow.com - id: 84d8d3-Y2U1Y PPT - SOLUTIONS PowerPoint presentation | free to view ... (Solutions) Mixtures a combination of two or more substances that do not combine chemically, but remain the same individual substances; can be separated by physical means Two types: Heterogeneous Homogeneous Creating Mixtures -

## Part 1 Procedures/Questions

Describe and draw what you see in the cups. Solutions - Science with Mr. Jones Selection of solutions resources. Including two ppts of core notes. Key words worksheet, HW work sheet and a couple of sheets relating to possible pracs and demos. Also a stepping stones frame work for students to use to plan a method for obtaining pure salt from rock salt. Solutions Resources | Teaching Resources In all solutions, whether gaseous, liquid, or solid, the substance present in the greatest amount is the solvent, and the substance or substances present in lesser amounts are the solute (s). The solute does not have to be in the same physical state as the solvent, but the physical state of the solvent

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