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replacement, or combustion
reaction. 1. $2\text{SO}_2 + \text{O}_2 \rightarrow 2\text{SO}_3$ 2.
 $\text{Al}_2(\text{SO}_4)_3 + 3\text{Ca}(\text{OH})_2 \rightarrow 2\text{Al}(\text{OH})_3 +$
 3CaSO_4 3. $2\text{C}_2\text{H}_2 + 5\text{O}_2 \rightarrow 4\text{CO}_2 +$

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reaction. 1. $2\text{SO}_2 + \text{O}_2 \rightarrow 2\text{SO}_3$ 2.

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Balancing Chemical Equations

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O combustion 10. Pb + FeSO 4

PbSO 4 + Fe single displacement

11. CaCO 3 CaO + CO 2

decomposition 12. P 4 + 3 O 2 2 P 2

O 3 synthesis 13. 2 RbNO 3 + BeF 2

... Types of Chemical Reaction

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AM ... Types of Chemical Reaction

Worksheet - #Chem@Work Atoms

are rearranged during chemical

reactions, and are not lost or

gained. Chemical reactions can be

represented using equations.

Catalysts speed up reactions

without being used up. Chemical

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thousands of common hazardous

chemicals, compatibility of

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chemistry classes and will actually be divided over several units:

Chemical Reactions (this unit)

Stoichiometry (next unit) Reaction

Energy (later units) Reaction Rates

and Equilibria (later units) This unit,

which is the first in that

list,... Chemical Reactions (Unit 6) |

Chemistry at HTHS Six Types of

Chemical Reaction Worksheet

Answers Balance the following

reactions and indicate which of the

six types of chemical reaction are

being represented: 1) $2 \text{NaBr} + 1$

$\text{Ca(OH)}_2 \rightarrow 1 \text{CaBr}_2 + 2 \text{NaOH}$ Type

of reaction: double displacement 2)

$2 \text{NH}_3 + 1 \text{H}_2\text{SO}_4 \rightarrow 1 (\text{NH}_4)_2\text{SO}_4$ 3)

$4 \text{C}_5\text{H}_9\text{O} + 27 \text{O}_2 \rightarrow 20 \text{CO}_2 + 18$

H_2O Type of reaction: synthesis

Type ... Six types of chemical

reaction worksheet -

MAFIADOC.COM Chemical Reaction

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HORROR LITERARY FICTION NON-

FICTION SCIENCE FICTION